

"Complex (soft) hydrodynamics"

From single paradigms to X-paradigm

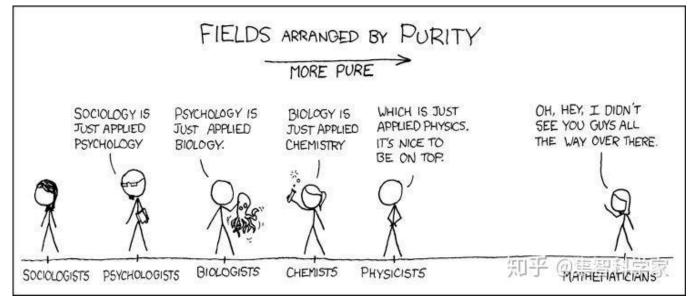


"MORE IS DIFFERENT"

P.W. Anderson (1923-2020, USA; 1977 Nobel-Phys)

"for their fundamental theoretical investigations of the electronic structure of magnetic and disordered systems" (shared with Mott, van Vleck)





Emergence (涌现论/演生论) v.s. Reductionism (还原论)

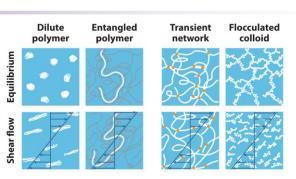
- 1. P.W. Anderson. More is different: broken symmetry and the nature of the hierarchical structure of science, Science, 1972
- 2. 段远源,专业基础课《高等工程热力学》,文献翻译作业

Emergence ... of what? What can be actually observed and measured?

When hydrodynamics meets "complex"

- What is *Hydrodynamics*? Why we need it?
 - quasi-particles as collective excitations

 - observables describing long-distance prop



- Dilute emulsion
- One way to develop a macroscopic theory with densities of physical quantities and the corresponding currents, is to combine
 - continuity equations (manifesting conservation laws), with
 - thermodynamic arguments ("constitutive relations"), between the macroscopic currents and the external bias to close the equations, to identify
 - how the entropy of the system responds to local density fluctuations of the conserved quantities
 - which requires the total entropy production rate to be non-negative
- They are **phenomenological** since they provide no means of calculating the coefficients in the constitutive relations, which is justified at distances that are much larger than any "microscopic" scattering **length scales**, the condition that is very often satisfied in experiments.
 - Narozhny, Hydrodynamics approach to 2D electron systems, 2022
 - 朗道,场论、流体动力学、物理动理学;谢多夫,连续介质力学
 - Ewoldt, Designing complex fluids, Annal Rev Fluid Mech, 2022

random

short-range interaction

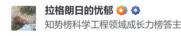
For "strongly-correlated" systems, does the hydrodynamic description still hold?

Contents



- From simple to complicated: find connections and differences
 - "Where to start ... with something new"
 - > "Decouple" charge correlation: learn from TWO figures
 - Personal experience A: Electrolyte transport at interfaces
 - Personal experience B: Transport of correlated electron systems
- Complex hydrodynamics as X-paradigm: What, Why, How, and Which?
 - Example: Emergence of quantum hydrodynamics
 - New physics: "More SCALES at INTERFACE is Different"
 - > Hallmark: Merging kinetic behaviors into hydrodynamics

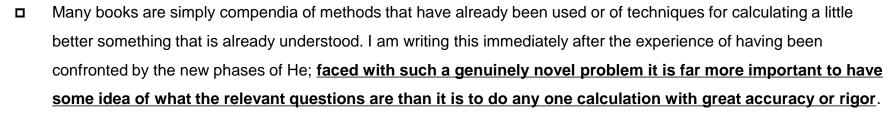
朗道的工作有多重要?



永公街的小魚等 205 人赞同了该回答

"Where to start ... with something new"

- P.W. Anderson (1923-2020, USA; 1977 Nobel-Phys)
 - □ I think most people entering research find that by far the most difficult question is where to start, especially when confronted with something that is actually new. This, ..., is the kind of question a book like this should be designed to answer.



- This is one of the reasons why I suggest that the two most important principles of condensed matter physics for our purposes are, first, **broken symmetry**, which tells us that what the **order parameter** is and what **symmetry** it **breaks** are the most vital questions; and, second, the **continuity principle**, which tells us to search for the **right simple problem** when **confronted with a complicated one**. To my way of thinking, detailed perturbation methods, and even Green's function and fluctuation-dissipation ideas, are somewhat less important, because they emphasize computation rather than understanding.
 - 1. Anderson. Basic notions of condensed matter physics, 1984.
 - 2. 朗道,统计物理学(I,Ⅱ)、物理动理学.

field convection? interface?
dynamics instability? pattern?
particle viscosity? vortex?

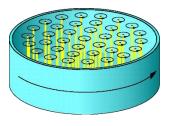
How to choose the <u>order parameter</u> and <u>adiabatic approximation</u>?

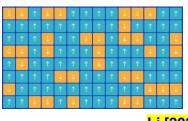
找到模式(及转换)的核心控制量[物理量(无量纲数)] 从熟悉的系统过渡[极限情况做起]

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Challenge I: Two typical correlated systems



Electrolytes in solutions

- Nature
 - √ dissolved
 - hydrodynamic
 - chemical

- Correlation
- ✓ steric
- ✓ electrostatic
- √ flow-induced

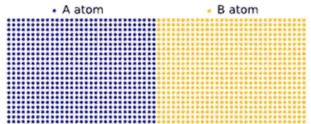
Electrons in solids

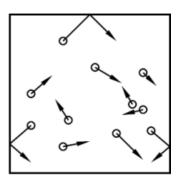
- Nature
- √ fermion statistics
- ✓ lattice regulation
- √ wave nature

- Correlation
- √ coherent
- ✓ electro- (e,l)²
- ✓ magneto- (↑,↓)



Familiar ones: rarefied gas transport (molecule)





short-range interaction

WHY diffusion decomposition of

ion transport in solution still stands?

- □ Advection
- □ Diffusion
- **□** Electro-migration

WHY particle kinetic description of

electron transport still stands?

- □ Excitation
- □ Propagation
- □ Scattering

Two figures: Onsager & Landau

Chen Ning YANG



L. Onsager (1903-1976, USA; 1968 Nobel-Chem)

- ES: electrolyte solution
- ES₁: liquid crystal theory
- ES2: dielectric theory
- ES₃: reaction-rate theory

- **IR: irreversible process**
- MF: Mathew function
- IR₁: fluctuation theory
- IR₂: turbulence statistics *****/

IS: 2D Ising model

IS₁: electron theory in metal

IS₂: vortex in superfluid He II

IS₃: superconducting in He II

Extra topics: ice's electrical theory, radiochemistry, molecular biology (life in the early days)

L.D. Landau (1908-1968, Russia; 1962 Nobel-Phys)

(QM) 量子力学的密度矩阵和统计物理学, 1927

(Mag) 自由电子抗磁性的理论, 1930

(Mag) 二级相变的研究, 1936-1937

(Mag) 铁磁性磁畴理论和反铁磁性解释, 1935

(CM) 超导体的混合态理论, 1934



1937, 原子核的几率理论 (PP)

1940-1941, 氦II超流性量子理论 (CM)

1954, 基本粒子的电荷约束理论 (PP)

1956, 费米液体的量子理论 (CM)

1957, 弱相互作用的 CP 不变性 (PP)



理论物理学教程 (朗道十卷)

The "Ten Commandments" (朗道十诫)

Contributions: Symmetry (ES-IR) & its breaking (IS; Mag-PP), correlation function (ES) & quasi-particle (CM-QM)

Onsager's theory on electrolyte in solution



WHY diffusion decomposition of ion transport in solution stands?

- Advection
- **Diffusion**
- Electromigration

Zur Theorie der Elektrolyte. I. Von Lars Onsager. § 1. Einleitung. Debye und Hückel haben vor einiger Zeit den Einfluß der interionischen Kräfte auf die thermodynamischen Eigenschaften sowie auf die Leitfähigkeit von Elektrolyten theoretisch berechnet1). Die Theorie der thermodynamischen Eigenschaften leistet nun gewissermaßen mehr als diejenige der Leitfähigkeit. Es wird nämlich z. B. für die Abhängigkeit der osmotischen Koeffizienten von der Konzentration der für hinreichend verdünnte Lösungen gültige Ausdruck: $g = 1 - \alpha \sqrt{c}$ abgeleitet, wobei im Koeffizienten a außer Tem-

1) Diese Zeitschrift 24 185, 305, 1923. Fig. 1. in einem gewisser Abstand vom Ion; wir be-zeichnen denselbe mit a, und nehmen ent-sprechend der z eiten Näherung von Debye

start from P. Debye's work

an; es bedeutet darin z den Abstand vom on, und es wird nach Debye und Hückel die von ihnen eingeführte, der Wurzel aus der Verdünnung proportionale mittlere Dicke der Ionenatmosphäre gleich - gesetzt. Wir erhalten dann, bei der äußeren Feldstärke E, für die Volum-

Total force balance, including electromigration & Stokes (viscous)

des Ions grenzenden Flüssigkeitsteilchen sollen dieselbe Geschwindigkeit haben. Die totale Kraft auf das Ion soll gleich & sein.

der Elektrolyte, I. Physik.Zeitschr.XXVII 1026

Die Gleichungen von Stokes lauten nun, wenn b = Geschwindigkeit, p = Druck, η = Viskosität der Flüssigkeit:

$$\eta \text{ rot rot } \mathfrak{v} = - \operatorname{grad} p + \mathfrak{F} \\
\operatorname{div} \mathfrak{v} = 0$$

Wir dürfen immer das Stromsystem in zwei

derart, daß v, dem Fall

R. -- R $\mathfrak{F}_1 = 0$

v2 dem Fall R2 - 0 1

> Viscous forces on the ion with superposition principle: apparent migration

electrophoretic

suchen das Stromsystem dv_2 , das von den zwischen den Abständen r und r + dr vom Ion angreifenden Kräften herrührt. Wir gelangen sofort zum Ziel, wenn wir die Resultate von Stokes über die Bewegung der Kugel heranziehen. Nach Stokes bewegt sich bekanntlich eine Kugel vom Radius r unter Einwirkung der Kraft P mit der Geschwindigkeit

durch die Flüssigkeit. Die pro Flächeneinheit von der Flüssigkeit auf die Kugel übertragene Kraft, die sich aus Druck und Reibungskraft zusammensetzt, ist dabei gleichmäßig über die Kugeloberfläche verteilt, und für alle Teile derselben der Bewegungsrichtung parallel und entgegengesetzt gerichtet1).

1) Vgl. Lamb, Hydrodynamics, Dritte Ausgabe S. 552.

- L. Onsager, Physikalische Zeitschrift, 1926.
- L. Onsager, J Chem Phys, 1931; Science, 1969.

12 December 1969, Volume 166, Number 3911

SCIENCE

The Motion of Ions: **Principles and Concepts**

"Anomalies" of strong electrolytes from the ideal additive behavior: Electrical conductivity,

Freezing point depression, Electromotive force

progress in our understanding of elec- action law and supported it by experi-

This article is the lecture he delivered in Sweden, when he received the Novel Prize in chemistry

ment at the first international congress in Karlsruhe, and within a few years Avogadro's principle gained wide ac-

Ra

We may at least speculate that contemporary developments in the kinetic theory of gases encouraged the chemists' change of attitude, although they rarely if ever admitted that; they preferred to maintain an inductive point of view in their publications. In 1860,

the van't Hoff "anomalies" and the electrical conductivity (1).

A greatly simplified picture trolyte solutions loomed. At fairly low but readily attainable concentrations solutions of readily dissociating compounds like hydrochloric acid, potassium hydroxide, and a great many salts like sodium chloride would be com-

pletely dissociated and the properties of a solution would be additive, not just over molecules, but even over the constituent ions. At higher concentrations, admittedly, one would have to allow for combination to form molecules or compound ions according to the mass-action law, as suggested by Ostwald (2). Nernst developed appropriate simple theories for the diffusion of electrolytes and for the variation of an electrode potential with the concentration of the ion discharged.

Such was the simple picture presented to me as a freshman chemist in 1920. In spite of some idealization it sufficed for a great many purposes; were eternally grateful for that. Howvery soon the journals rather

Whether the experimenters studied the electrical conductivities or the equilib rium properties like freezing point de pressions and electromotive forces, sign nificant deviations from the ideal ad trolytes since the days of Arrhenius, ments. After a while (1885) van't Hoff ditive behavior persisted to much lowe concentrations than had been predicte

according to the mass-action law from

the measurements performed on more

- concentrated solutions. These phonom ena became known as the "anomalies" of strong electrolytes. In many ways the anomalies displayed conspicuous regularities; if one compared salts of the same valence type like NaCl and KNO3, the differences were typically small even at concentrations as high as 0.1 mole/liter. Suspicion centered on the long-range electrostatic forces

Debye and Hückel finally succeeded

"for the discovery of the reciprocal relations bearing his name, which are fundamental for the thermodynamics of irreversible processes"

Landau's theory on electron evolution in solid



Fermi liquid

 $\eta = (v_F^{\eta})^2 \tau_{ee,2}/4$

Landau Fermi-liquid

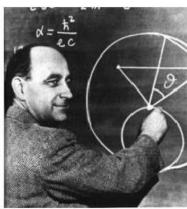
From Fermi gas to Fermi liquid

WHY

particle kinetic
description of
electron transport
still stands?

- □ Excitation
- □ Propagation
- Scattering



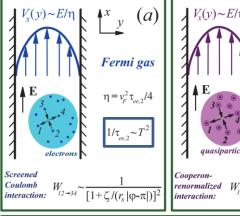


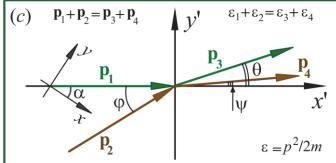
In 1956, Landau developed a theory of interacting spin-1/2 fermions¹

The Landau-Fermi liquid theory successfully describes metals, nuclear matter, liquid He-3 ...

At low temperatures, the average excitation energy is $\sim k_BT$







- 1. L. D. Landau, The theory of Fermi liquids, Zh Eskp Teor Fiz, 1956
- 2. P.S. Alekseev & A.P. Dmitriev, Viscosity of two-dimensional electrons, *Phys Rev B*, 2020

"for his pioneering theories for condensed matter, especially liquid helium"

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Challenge II: From familiar to unfamiliar ones

Correlation

steric

electrostatic

flow-induced



Electrolytes around interfaces

From bulk, solid to liquid interface

- Nature dissolved
 - hydrodynamic

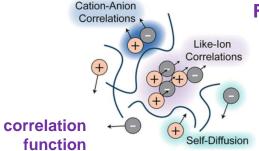
 - chemical

Quantums in solids

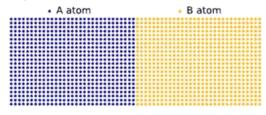
From phonon to electron

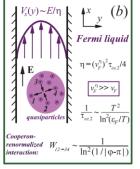
- Nature
 - boson/fermion
 - lattice regulation
 - wave nature

- Correlation
- coherent
- electro- (e,l)2
- magneto- (↑,↓)



Familiar ones - I: nature and transport behavior of electrolytes in solution and electrons in solid





quasi-particle picture

Familiar ones – II: transport of electrolyte transport at solid interface & phonon transport in solid

WHETHER kinetic description of ion

transport at solid interface still stands?

- Nature of interface/phases
- **Adsorption kinetics**
- *lon transport dynamics*

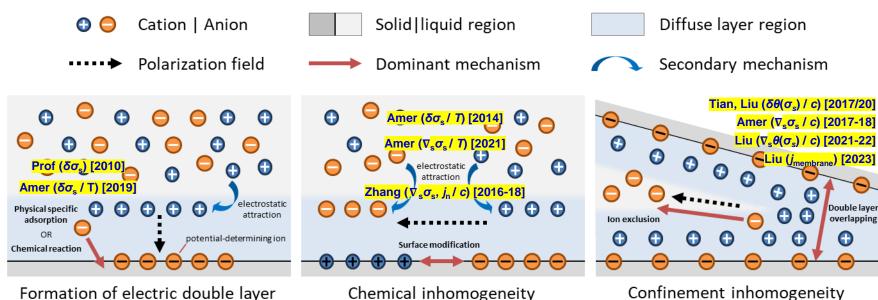
WHETHER kinetic description of

phonon transport still stands?

- Total no. conservation
- Frequency domain width
- Inter-quasi-particle scattering

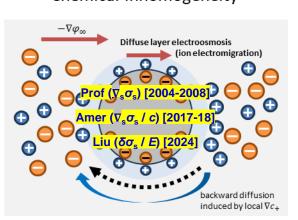
Problem A: Electrolyte transport at interface (S)



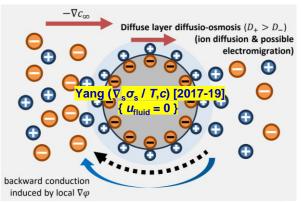


Diffuse layer streaming (ion convection) Backward conduction backward diffusion induced by local $\nabla \varphi$ induced by local ∇c_+

Pressure gradient upon an enclosure



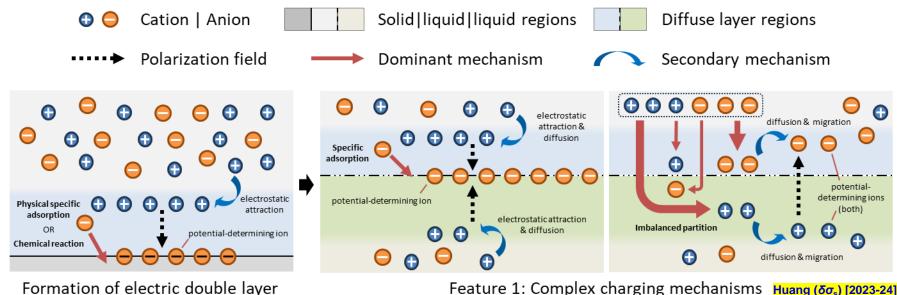
Confinement inhomogeneity



Electric field upon an enclosure Concentration gradient upon an enclosure

Challenge: Electrolyte transport at interface (L)





Formation of electric double layer

Diffuse layer streaming

(ion convection)

backward diffusion

induced by local ∇c_+

Enhanced charge adsorption Enhanced distribution potential through interface slip across liquid-liquid interface Diffuse layer streaming Diffuse layer streaming backward diffusion backward diffusion 0 O backward diffusion backward diffusion

Pressure gradient upon an enclosure

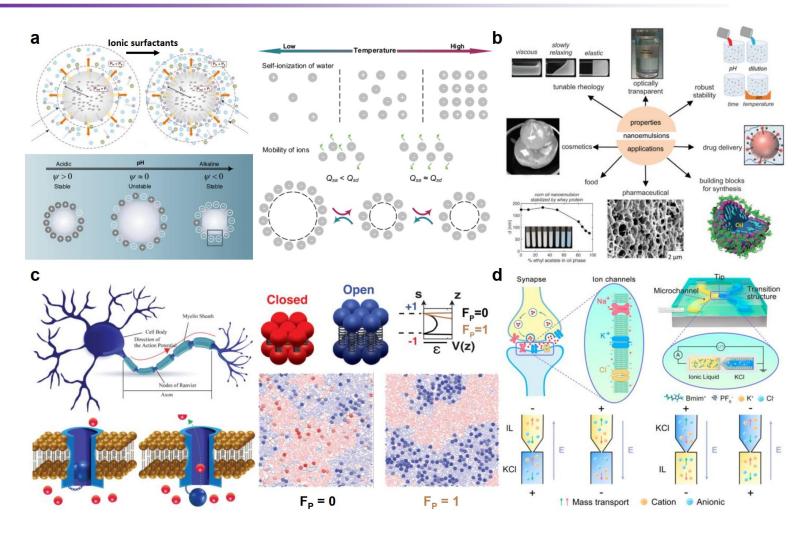
Backward conduction

induced by local $\nabla \varphi$

Feature 2: Slip-induced inhomogeneous charging

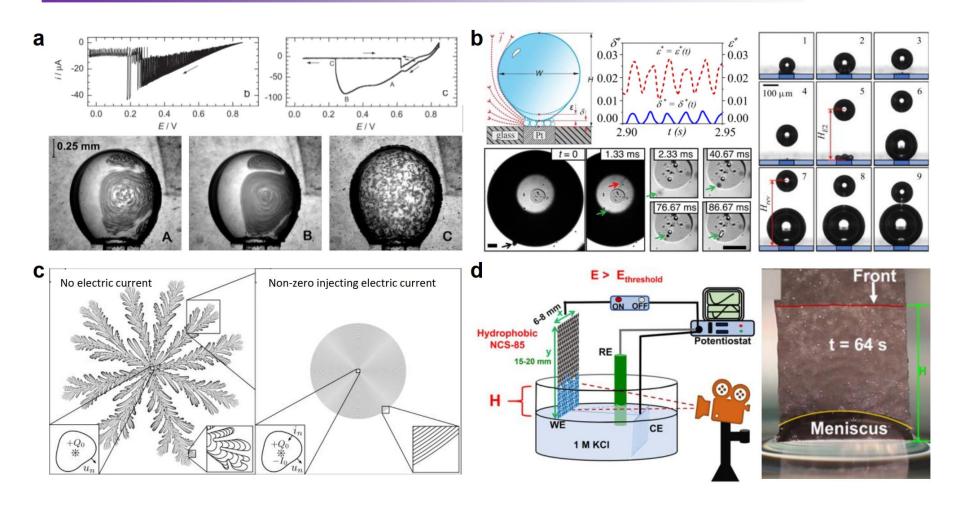


Electrokinetic multiphase flow: interface charging



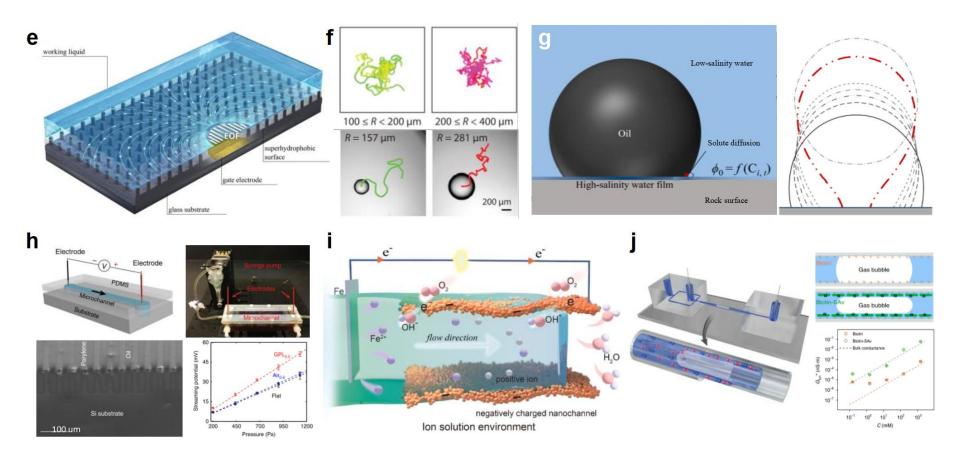
- a) Li et al., JCIS, 2021; Ma et al., JCIS, 2022 / 2024.
- b) Gupta et al., Soft Matter, 2016.
- c) Torbati et al., Rev Mod Phys, 2022; Suma et al., PRX Life, 2024.
- d) Li et al., Nano Lett, 2024.

Electrokinetic multiphase flow: field-driven flow



- a) Trojanek et al., Electrochem Comm, 2017.
- b) Bashkatov et al., Phys Rev Lett / Phys Chem Chem Phys / Journal of Fluid Mech, 2019/2022/2023/2024.
- c) Mirzadeh et al., Phys Rev Lett, 2017.
- d) Pan et al., 2023.

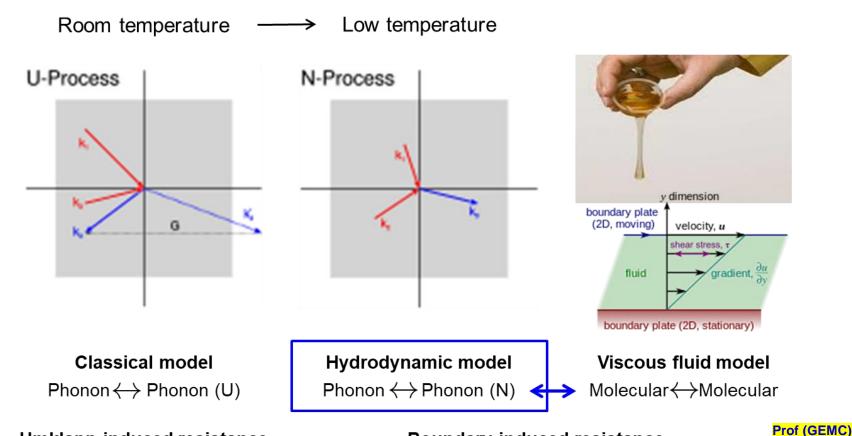
Electrokinetic multiphase flow: ion-mediated transport



- e) Dehe et al, Phys Rev Fluids, 2020.
- f) Suda et al, Phys Rev Lett, 2021; Michelin, Annal Rev Fluid Mech, 2023.
- g) An et al, Fuel, 2022.
- h) Fan et al, Nat Comm, 2018.
- i) Li et al, Nano Energy, 2021.
- j) Ma et al, Nat Comm, 2020.

Problem B: Transport in quantum systems (ph)





Umklapp-induced resistance

Boundary-induced resistance

[99-24]

(Fourier law)

(Hydrodynamic flow)

Guo (Meso & Macro) [2013-18]; Miao, Ran (Meso) [2016-22]; Liu (Macro/Micro) [2023-24]

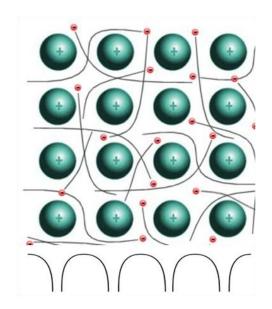
Phonon hydrodynamics!

Challenge: Transport in quantum systems (e)



Room/ultralow temperature

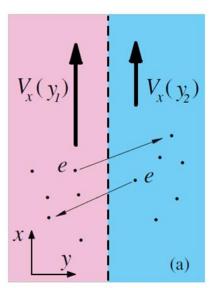
- dimension & ultrapu
- Low temperature & low Metal, Si|Ga[Al]As, Doped-S|BLG, Pt|PdCoO₂
 - dimension & ultrapure SLG, semi-metal (MoP, WP2), topo-insulator

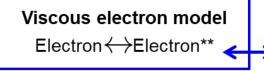


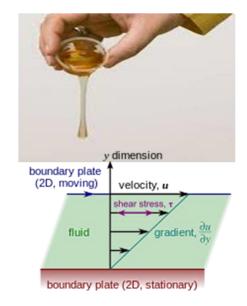
Nearly Free electron model
Electron ← Lattice*

Lattice-induced resistance (Ohmic flow)

Miao (Meso) [2017-22]







Viscous fluid model Molecular↔Molecular

Boundary-induced resistance

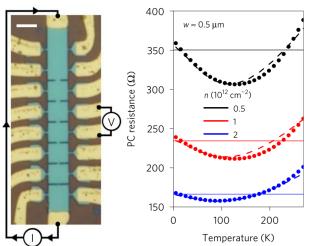
(Hydrodynamic flow)

Huang (Meso & Macro) [2017-21]; Meng (Macro) [2019]

Electron hydrodynamics!

Electron hydrodynamics: transit – σ_e





Super-ballistic flow in confined graphene

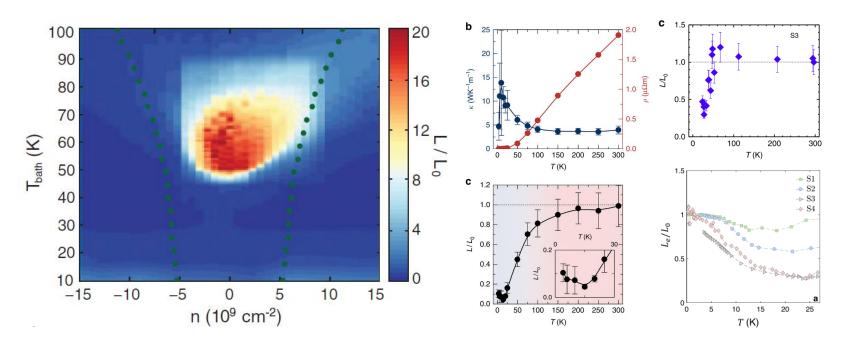
High electron conductivity

Low heat generation → High current transport capability

High cutoff frequency → **Low** switching time of the transistor

- ✓ System: 2D materials, ultra-pure, microscale and low-T
- ✓ Phenomena: high mobility, may improve performance of semiconductors

Electron hydrodynamics: energy – ZT_e



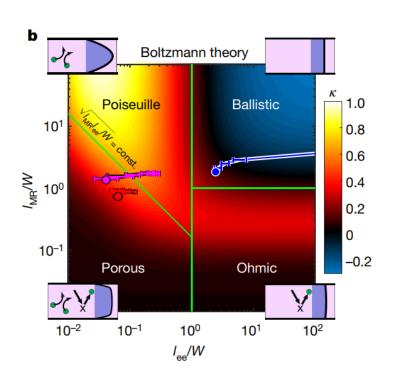
Breakdown of W-F law in graphene (left) & WP₂ (mid) & MoP (r-top) & Sb (r-bot)

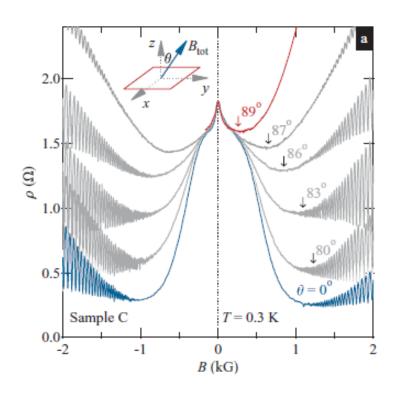
$$ZT = S^2/\mathcal{L}$$
 $\mathcal{L} \equiv rac{\kappa_{
m e}}{\sigma T} = rac{\pi^2}{3} \Big(rac{k_{
m B}}{e}\Big)^2 \equiv \mathcal{L}_0$

higher/lower $\kappa_{\rm e}$ than W-F law

- ✓ System: 2D material, ultrapure, low T
- ✓ Phenomena: Anomalous transport of 2D electrons
- ✓ Impact: low heat loss in thermoelectric materials
- 1. Crossno J. et al. Science, 351: 1058-1061, 2016.
- 2. Gooth, J., et al., Nat Comm. 9(1): 4093, 2018.
- 3. Jaoui, A., et al., npj Quantum Materials, 3(1): 64, 2018.
- 4. Kumar, N., et al., Nat Comm, 10(1): 2475, 2019.
- 5. Jaoui, A., B. Fauqué, and K. Behnia, *Nat Comm*, **12**(1): 195, 2021.

Electron hydrodynamics: info – $\sigma_e(B,T)$





Conduction regime with different mechanisms (confinement)

Giant negative magneto-resistance in GaAs/AlGaAs (magnetic field)

Magnetic field will dramatically impact the electric resistance in confined materials.

- 1. Hatke A.T. et al. Phys Rev B, 85: 081304, 2012.
- 2. Alekseev P. S. and Semina M. A. Phys Rev Lett, 98: 165412, 2018.
- 3. Chandra M, Kataria G, Sahdev D, et al. Phys Rev B, 99: 165409, 2019.
- 4. Mandal I, Lucas A. Phys Rev B, 101(4): 045122, 2020.

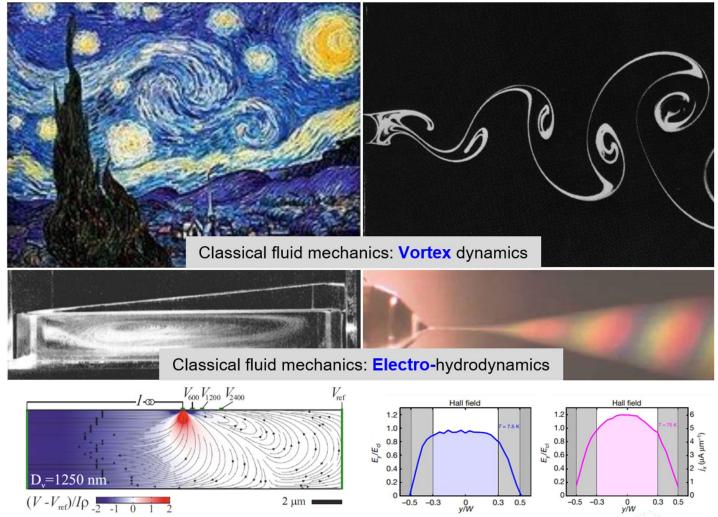
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From one paradigm (electro-hydrodynamics)





Novel phenomena in 2D materials: Electron backflow | From Ballistic to Viscous regime

And another paradigm (physical kinetics) ...



Boltzmann transport equation (BTE) – semi-classical description

collective moving velocity

$$\frac{\mathrm{d}f}{\mathrm{d}t} = \frac{\partial f}{\partial t} + \boldsymbol{v} \cdot \frac{\partial f}{\partial \boldsymbol{r}} + \frac{\boldsymbol{F}}{m} \cdot \frac{\partial f}{\partial \boldsymbol{v}} = \boxed{C(f)} \begin{cases} \text{Resistive scattering} & f_0(\boldsymbol{r}, \boldsymbol{\varepsilon_k}) \\ \text{Conservative scattering} & f_0(\boldsymbol{r}, \boldsymbol{\varepsilon_k} - \boldsymbol{u} \cdot \boldsymbol{p}) \end{cases}$$

Change of particle state Particle Different mechanisms Different (quasi-) equilibrium states

f: the non-equilibrium distribution function of the particle cluster around (r, v)

Nanoscale Energy Transport and Conversion:
A Parallel Treatment of Electrons, Molecules, Phonons, and Photons

纳米尺度能量输运和转换:
对电子、分子、声子和光子的统一处理
[美] Gang Chen 著

Solving BTE: Deterministic | Stochastic (particle nature)
Upscaling BTE: Hydrodynamic description (macroscopic)

Beyond BTE: wave nature, strong correlation, scattering rate

<u>Coherence</u>: Quantum transport in low-D system

<u>Localization</u>: Strong disordered/correlated system

<u>Super/Magneto/Topo</u>: Strong (spin/Coulomb) correlated system

- 1. Rammer. Quantum transport theory. Perseus, 1998.
- 2. Datta. Quantum transport Atom to transistor. Cambridge University Press, 2005
- 3. Nazarov. Quantum transport Introduction to nanoscience. Cambridge University Press. 2009.
- 4. G. Chen. Nanoscale energy transport and conversion, Tsinghua University Press, 2014

Into X-paradigm (quantum hydrodynamics) ...



Boltzmann transport equation (BTE) – Chapman-Enskog expansion

$$\frac{\mathrm{d}f}{\mathrm{d}t} = \frac{\partial f}{\partial t} + \boldsymbol{v} \cdot \frac{\partial f}{\partial \boldsymbol{r}} + \frac{\boldsymbol{F}}{m} \cdot \frac{\partial f}{\partial \boldsymbol{v}} = C(f) \qquad f = f^{(0)} + \epsilon f^{(1)} + \epsilon^2 f^{(2)} + \dots$$

Classical N-S equation

$$\frac{\mathrm{D}\boldsymbol{P}}{\mathrm{D}t} + \nabla \cdot \boldsymbol{\sigma} = \rho \boldsymbol{g}$$

$$\boldsymbol{\sigma} = p\boldsymbol{I} - \mu \left(\frac{2}{3} (\nabla \cdot \boldsymbol{u}) \boldsymbol{I} + \nabla \boldsymbol{u} + (\nabla \boldsymbol{u})^T \right)$$

Phonon N-S equation

$$\left| \frac{\partial \boldsymbol{q}}{\partial t} \right| + \nabla \cdot \boldsymbol{Q} = -\frac{\boldsymbol{q}}{\tau_{\mathrm{R}}}$$

$$\mathbf{Q} = \frac{1}{3}v_{g}^{2}e\mathbf{I} - \frac{1}{5}v_{g}^{2}\tau_{N}((\nabla \cdot \mathbf{q})\mathbf{I} + \nabla \mathbf{q} + (\nabla \mathbf{q})^{T})$$

Electron N-S equation

$$\frac{\partial \mathbf{P}}{\partial t} + \nabla \cdot \mathbf{T} = -\frac{\mathbf{P}}{\tau_{\text{MR}}} + m^* \mathbf{F}_{\text{macro}} \mathbf{T} = \mathcal{P} \mathbf{I} - \mu_{\text{e}} ((\nabla \cdot \mathbf{u}) \mathbf{I} + \nabla \mathbf{u} + (\nabla \mathbf{u})^T)$$

- 1. Y.Y. Guo. 非傅里叶导热的宏观声子输运模型及非平衡热力学. 博士学位论文, 2018.
- 2. Y.F. Huang. 微纳尺度低维电子输运的水动力学研究. 本科毕业论文, 2019.

"More SCALES at INTERFACE is Different"



Electrolytes around interfaces

From solid to liquid

- Mean distance (background)
 - $n_{\infty}d_{\rm M}^3 \sim 1$
- Radius of ion ("coherence")
 - radius variation hydration
- Debye length (screening)
 - $\qquad \varepsilon (\delta \varphi / \lambda_{\rm D})^2 \sim (\delta n_{\rm \infty}) \ k_{\rm B} T$
 - concentration ratio
 - solvent permittivity ratio
- Bjerrum length (correlated)
 - $k_{\rm B}T \sim e^2/\varepsilon d_{\rm B}$
 - solvent permittivity ratio
- Relaxation time ("scattering")
 - $\tau_{\rm diff} \sim \lambda_{\rm D}^2 / (k_{\rm B} T / \mu_{\rm m})$
 - frictional force $[\mu, E, ...]$
 - viscosity ratio (gas/self?)
 - dissociation constant (weak?)

Quantums in solids

From phonon to electron

$$[I_{tri}], I_N, I_R, W, [I_{dis}] \longrightarrow \begin{bmatrix} \lambda_D \end{bmatrix}, I_N, I_R, W, \begin{bmatrix} I_{dis} \end{bmatrix}$$

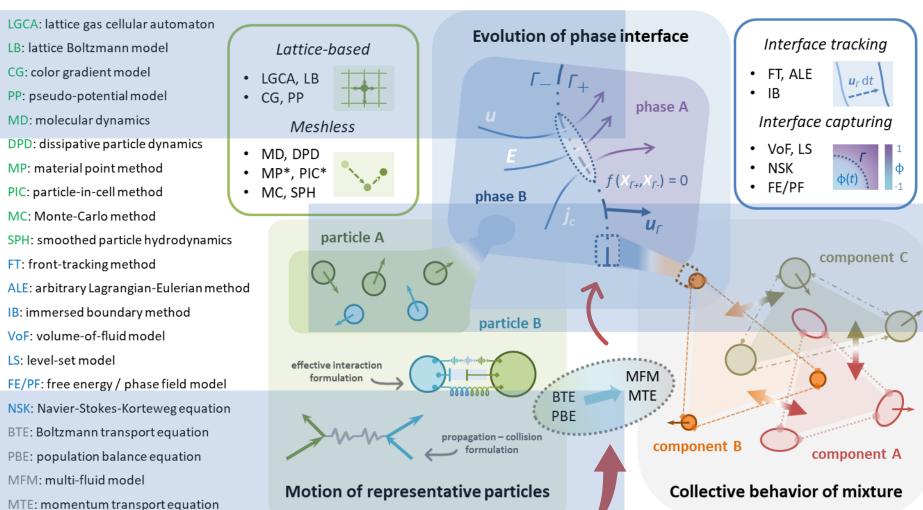
 $[E_F, U_{corr}, \beta_{E,B}]$

- Lattice periodicity (background)
 - interaction energy uniformity
 - steric effect (entropy)
- Energy wavepacket (coherence)
 - wave vector [volume] Fermi surface
 - frequency [life] Fermi energy
- Debye length (screening)
 - $\qquad \varepsilon \, (\delta \varphi / \lambda_D)^2 \sim (\delta n_{\scriptscriptstyle \infty}) \, k_{\rm B} T$
 - concentration prescribed
 - lattice permittivity
- Bjerrum length (correlated)
 - $k_{\rm B}T \sim e^2/\varepsilon d_{\rm B}$
 - lattice permittivity correlation
- Relaxation time (scattering)
 - $1/\tau_i \sim \Sigma_{n,\Omega} |\Delta v| |d\sigma/d\Omega| (f_1f_2 f_1'f_2')$
 - scattering events: X-X (N/U), X-Y

General Nature (physics)

"Interface" from the perspective of multiscale ...





新物理源自新研究对象(视角): 新界面 → 新尺度 → 新机理 → 新描述 → 新方法

X-paradigm: What, Why, How, and Which?



- 何为力学 (工科基础) 相关的交叉学科?
 - 《科学革命的范式》范式及其转换
 - 《论技术科学》例外来自/创造工程需求
 - 交叉特征: 界面接触 + 内核融合 + 进化独立
 - 现实投影: 工业需求 + 大牛烙饼 + 众人推广
 - 典型案例: 软物质物理, 计算神经科学
- 原学科范式 力学 / 物理 / 化学
 - 经典[电]流体力学: [electro-]hydrodynamics
 - ✓ 电化学/胶体科学 (离子吸附): chemical kinetics
 - ✓ 与/或,凝聚态物理 (粒子散射): physical kinetics

电动多相流体力学 量子水动力学

a) 出现新"界面":带电液液界面 | 守恒散射截面

b) 涌现新"尺度": 介电常数比 | 守恒散射率

c) 新"研究对象":电动双侧耦合 | 量子集体运动

d) 引入新"机理": 界面极化 | 量子粘性

e) 要求新"描述": 有效边界 | 水动力学

f) 亟需新"方法": 摄动展开 | 升尺度展开

- 新学科范式 或仅作为名词归纳
 - 复杂流体力学: complex (soft) hydrodynamics
 - 研究对象: 含内部额外(强关联)自由度的流体体系
 - 复杂机理举例:奇界面,多物理,子结构 "序参数"
 - □ 子分支 1: multiphase electrokinetic hydrodynamics
 - □ 子分支 2: quantum hydrodynamics
 - □ 子分支 3: soft flowing matter physics (micro-rheology)
 - 关键图像 merge kinetics into hydrodynamics
 - 传统手段: <u>物理实验 → 理论模型</u> → 数值模拟(实验)
 - 新兴手段:理论建构 与 数据处理 的结合

多尺度系统的对策与挑战

┏ 一奇界面:摄动理论-时空分区,参数要求高、难延拓

→子结构: 粗粒理论-代表单元, 大尺度模型、难整合

▶ 多物理:有效理论-机理提取,时空强关联、难解耦

▶反问题:低维理论-去除冗余,优化成本高、难规划

Last but not at least ...



- From simple to complicated: find connections and differences
 - "Where to start ... with something new"
 - > "Decouple" charge correlation: learn from TWO figures
 - > PhD proposal: Electrolyte transport at two-liquid interfaces
 - Undergrad thesis: Transport in weakly-correlated electron systems
- Complex hydrodynamics as X-paradigm: What, Why, How, and Which?
 - **Example:** Birth of Huang's Undergrad/PhD proposals
 - > New physics: more "insights beyond local consensus" is different
 - ▶ Hallmark: The first paper (易上手 避免强耦合, 抓本质 敢于做假设)
 - ▶ Positive feedback requiring: 良师益友,敢想敢干,博观约取,厚积薄发



Thank you